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# Uncertainty analysis of underwater hyperspectral attenuation measurement

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The uncertainty of the underwater hyperspectral attenuation measuring instrument (ACS) is studied. Through the Mie scattering theory of standard particles of different sizes (2, 5, 10, 20 μm) On the comparison between the calculated value and the measured value of the ultraviolet-visible spectrophotometer (PE35), it is concluded that the maximum measurement error of the attenuation of PE35 does not exceed 8%. Against c In the turbid water environment, ACS and PE35 were used to measure the attenuation of turbid seawater samples in the East China Sea. The results show that: ACS measurement under turbid water The results are underestimated, and the uncertainty is negatively correlated with the wavelength; the turbidity of the water body has a great influence on the uncertainty of the ACS attenuation measurement, and it is positively correlated. The measured value of ACS (10 cm) is underestimated by 17.2% ~ 19.04% under high water, and the measured value of ACS (25 cm) is underestimated by 7.84% ~ 15.36% under high turbidity The underestimation of ACS (10 cm) increased to 26.4% ~ 28.24%.

Keywords marine optics; attenuation coefficient; Mie scattering calculation; uncertainty  
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## Analysis of Uncertainties Associated to Underwater Hyperspectral Attenuation Measurements

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Abstract The uncertainty for the underwater hyperspectral measurement of attenuation (ACS) is studied. The attenuation measurements for standard particles of different diameters (2,5,10,20 μm) by ultra violet-visible (UV-Vis) spectrophotometer (PE35) are compared with model values by Mie scattering theory, and the attenuation measurement error of PE35 is found less than 8%. Subsequently, simultaneous attenuation measurements by ACS and PE35 are carried out for samples from the high turbid water in the East China Sea. Comparative results show the attenuation measurements by ACS are underestimated in high turbid water, and the uncertainty of ACS measurements has a negative correlation with wavelength, conversely, with a strong positive correlation with turbidity of water. The attenuation measurements of ACS (10 cm) and ACS (25 cm) are underestimated within 17.2% ~ 19.04% and 7.84% ~ 15.36% in low turbid water, respectively, while ACS (10 cm) is within 26.4% ~ 28.24% in high turbid water.

Key words oceanic optics; attenuation coefficient; Mie scattering calculation; uncertainty  
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The spectral attenuation coefficient  $c(\lambda)$  of the water body refers to the attenuation rate of light per unit distance of each wave band in the water body. The light attenuation of natural water is main The absorption and scattering compositions of dissolved organic matter and suspended particles have the following relationship with the absorption coefficient  $a(\lambda)$  and the scattering coefficient  $b(\lambda)$ :  $c$  In the measurement of the actual optical volume (IOP) of water, due to the lack of accurate experimental measurement methods for  $b(\lambda)$  measurement, the direct measurement accuracy is low, which is u

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0401004- 1

The difference between  $c(\lambda)$  and  $a(\lambda)$  to estimate the total scattering coefficient  $b(\lambda)$  of the water body. Therefore, in the actual research process, the attenuation coefficient  $c(\lambda)$  and the absorption c The radiation coefficient  $b(\lambda)$  jointly becomes the radiance  $L(\lambda)$  of the water body, which is the three most basic inherent optical quantities of this important water color remote sensing physical parameter [1-3]. by

Since  $a(\lambda)$  and  $b(\lambda)$  are both variable functions of the nature of natural water and the concentration of each component, so  $c(\lambda)$  comprehensively reflects that light is suspended in water. The degree of influence of particles, soluble organic matter and plankton is a direct indicator of factors such as the transparency of water bodies and the concentration of suspended solids. Many scholars have studied the correlation between the attenuation coefficient of natural water and organic particulates, inorganic particulates, chlorophyll and other parameters. There is a clear correlation between the near-shore turbid water body with high suspended solids concentration and the inland lake [9]. Philips et al. [7] on the coast of Florida, USA. Linear regression analysis of the water attenuation coefficient of the water body and inorganic particles, chlorophyll a and water color found that the correlation between the attenuation coefficient and inorganic particles is very good. Pak et al. [8] found that the particle beam attenuation coefficient  $c(660\text{ nm})$  determined by the nature of suspended particles in water has a high correlation with the total particle volume concentration. Guan Wenjiang et al. [9] found that the chlorophyll concentration data and the optical characteristics of pure seawater can be used to calculate the attenuation coefficient of the water body in the near-shore. Zhang Yunlin et al. [10] found that the increase of suspended solids concentration in Taihu Lake is the main reason for the decrease of the transparency of the water and the increase of the optical attenuation. And establish the power function relationship between the optical attenuation coefficient and the concentration of suspended matter.

At present, the measurement of  $c(\lambda)$  can be obtained through laboratory measurement and on-site measurement. The commonly used measuring instrument in the laboratory is a spectrophotometer. After the sample is received by the spectrophotometer optical receiver, the attenuated light contains part of the forward scattered light, so the measurement error of the attenuation coefficient is mainly caused by forward scattering. The measurement accuracy of forward scattering is on [11-12]. In order to reduce the measurement error, the receiving angle of the beam receiver should be smaller than the divergence angle of the light source. Etc. [13-14]. When the attenuation coefficient is measured using a spectrophotometer from the sample placed in a position closer to the source, and increasing the distance between the sample and the beam receiver. At the same time, a special slit baffle is placed between the sample and the light receiver to reduce the interference of forward scattering in the transmitted light and improve the measurement accuracy. Shi et al. [15] used Bricaud's method to verify the measurement method using standard particulate matter (center particle size  $1.999\ \mu\text{m}$ ) when measuring the attenuation coefficient of algae. Reliability. Compared with the spectrophotometer, the hyperspectral attenuation measuring instrument (ACS) that can be applied on-site underwater needs to have rigidity with good airtightness. The collimated light path [16] to adapt to high-pressure underwater environment. Underwater 9-band and hyper-spectral absorption attenuation instrument (AC-9 and ACS) is currently the most widely used underwater attenuation measurement instrument in the marine field. However, the previous analysis of the uncertainty of the instrument measurement is more limited. There are few studies. It has been considered that the measurement error of AC-9 attenuation is mainly due to the angle of view of the beam receiver (0.9), which is larger than the photon passing through the distribution angle leads to an underestimation of  $c(\lambda)$  [17-19]. Zaneveld et al. [20] proposed a "ratio method" to correct the error. Leymarie et al. [21] believed that the increase in the optical path of the photon caused by single and multiple scattering during the photon propagation and the tube wall is also the source of the  $c(\lambda)$  measurement error. The Monte Carlo model is used to simulate and analyze the measurement accuracy of ACS. The results show that  $c(\lambda)$  in complex water bodies is usually underestimated by 10% to 40%. Most scholars use the method of numerical simulation to analyze the uncertainty of the measurement of the underwater hyperspectral attenuation instrument, and analyze it with the measured data. And the verification method has not been reported. In recent years, large amounts of underwater high spectral attenuation instrument, so it is very necessary to evaluate the uncertainty of ACS measurement in China's highly turbid water.

In order to simulate the influence of particles with different forward scattering intensities on the attenuation measurement of the spectrophotometer, this paper uses 4 different standard particle sizes (5, 10, 20  $\mu\text{m}$ ) Mie scattering theory calculated values and measured values to comprehensively evaluate the uncertainty of spectrophotometer measurement. Using ACS-based laboratory circulation measurement system and a spectrophotometer were built to simultaneously measure the water samples of the Yangtze River estuary in July 2012, and ACS actual measurements were compared. The more accurate uncertainty range under the conditions has important reference value for the accuracy evaluation of ACS attenuation measurement in high turbid waters in China.

## 2 Materials and methods

### 2.1 Selection of experimental samples

The experimental samples were selected from 4 different sizes of Thermo's Duke Standards™. Microsphere Size Standards 4000 series, standard quasi-particulate matter is mainly composed of polystyrene, and the central particle diameters are 1.999, 4.993, 10.12, and 19.99  $\mu\text{m}$ , respectively, and the central particle diameter distribution conforms to the standard errors are 0.022, 0.05, 0.09, and 0.28  $\mu\text{m}$ . Since polystyrene has no absorption characteristics, only the refractive index actual value is considered in the calculation. Pass the particle size and distribution of particles, the number and concentration of particles in the suspension, the refractive index of water and standard particles. The Mie scattering theory can be calculated to calculate its attenuation coefficient at the same concentration (where the calculation code of Mie scattering uses Christian Code published by Mätzler [22]). Comparing the calculated value of the standard particulate matter and the measured value of the spectrophotometer at the same concentration can determine the spectrophotometer uncertainty.

The measured samples were selected from the July 2012 Yangtze River estuary seawater samples. After retrieval, they were evenly mixed using a magnetic stirrer. Different turbidity and total suspended matter. There are 8 groups of samples (number: Sample\_a ~ Sample\_h) of float concentration (TSM), as shown in Table 1.

0401004- 2

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Table 1 Seawater samples from the Yangtze Estuary in July 2012  
Table 1 Seawater samples of Changjiang River in July 2012

Sample	Sample_a	Sample_b	Sample_c	Sample_d	Sample_e	Sample_f	Sample_g	Sample_h
Turbidity / NTU	twenty three	104.5	265.1	381.3	551.2	662.8	705.4	886.4
TSM / (mg / L)	9.81	53.18	325.65	190.72	424.24	544.35	616.3	660.88

### 2.2 Measurement of attenuation coefficient by spectrophotometer

The spectrophotometer uses the UV-Vis spectrophotometer PE35 produced by Perkin Elmer, with a wavelength range of 190 ~ 1100 nm, and stray light is less than 0.01% ( $T$  is the transmittance), the resolution is 0.1 nm, and the spectral bandwidth is 1 nm. There are many factors that can affect the process of laboratory attenuation measurement. The accuracy of the measurement results, such as the polarization and instability of the light source, stray light, photometric linearity, etc. [23-25], place the sample close to the light source during measurement. Increase the distance between the sample and the beam detector, and add a special black round baffle to the detector to reduce the impact of forward scattering on the measurement results (connect the closing angle is reduced to 0.3°). The measurement uses a 1 cm cuvette, Milli-Q standard pure water as a reference, and the attenuation coefficient is

$$c(\lambda) = \frac{2.303}{l} [D_s(\lambda) - D_r(\lambda)] \quad (1)$$

Where  $l = 1\text{ cm}$  is the optical path of the cuvette,  $D_s(\lambda)$  is the absorbance of the sample,  $D_r(\lambda)$  is the absorbance of Milli-Q standard pure water and blank cuvette. sea water samples also use this method to measure their attenuation coefficient.

### 2.3 ACS measurement of water attenuation coefficient

The underwater hyperspectral absorption attenuation measuring instrument produced by the Wetlab company in the United States is divided into two models according to the length of the measurer ACS (10 cm) and 25 cm optical path ACS (25 cm). There are 85 spectral channels in the visible band, with a spectral resolution of 4 nm and a measurement accuracy of 0.01 m<sup>-1</sup>. Before measuring the sample, the instrument performs pure water correction and temperature and salt correction in sequence, and finally obtains the attenuation coefficient of the water body con

$$c_m(\lambda) = c_{raw}(\lambda) - c_{off}(\lambda), \tag{2}$$

$$c_{m,TS} = c_m [\psi_T(T - T_r) + \psi_S(S - S_r)], \tag{3}$$

Where  $c$  is the attenuation coefficient,  $\lambda$  is the wavelength, the subscript  $m_{raw}$  represents the original measurement data,  $m$  represents the measurement data,  $off$  represents the offset of the pure water correction. The reference value is shown, and the subscript  $m_{TS}$  indicates the temperature and salt corrected data,  $T$  is the temperature,  $S$  is the salinity, and  $\psi_T$  and  $\psi_S$  are the temperature and salinity correction coefficient. Since the attenuation coefficient of pure water is a constant, this article only compares and analyzes the attenuation coefficients of water components except pure water.

However, the traditional method of using ACS is on-site underwater measurement, and a large number of seawater samples are required in the laboratory according to the underwater measurement. To increase the difficulty and cost of the measurement experiment, in order to solve this problem, an ACS flow cell measurement system that can be used for laboratory measurement was built. As shown in Figure 1. Connect the cleaned Teflon water pipes to the ACS (25 cm) and ACS (10 cm) attenuation measurement tubes (hereinafter referred to as the  $c$  tube, in order to avoid The black rubber tube is used at the optical interface), the water pipe on the left side of the ACS instrument is connected to the three-way valve, and the interface is connected to the calibration cylinder, with Flat surface, used for calibration, the other interface is connected to the peristaltic pump, used for circulating system circulation power, the third interface is connected to the buffer pool, and finally connect The  $c$  tubes of ACS (10 cm) and ACS (25 cm) form a closed loop measurement system. In order to solve the internal influence of the existence of bubbles in the circulation system

Figure 1 ACS laboratory circulation measurement system  
Fig.1 ACS laboratory cycle measurement system

0401004- 3

Light learn learn Report

The problem of the measurement result shall be sealed at all interfaces. Adjust the peristaltic pump to the proper pressure to make the water in the water pipe flow slowly before the formal measurement. The bubbles in the water are flushed into the buffer tank. If the fluctuation of the curve measured by the ACS is stable within 0.005 m<sup>-1</sup>, it is considered that the bubbles in the system have been eliminated. When using, you need to use the Milli-Q water that has been left in the calibration cylinder for one day to remove bubbles. The entire measurement system is slowly filled. After the bubbles are removed, The temperature of the water in the pool. After the calibration is completed, turn off the water flow switch of the calibration cylinder to make Milli-Q water flow slowly in the system. When measuring seawater, Slowly add seawater samples into the pool and turn on the switch on the side of the buffer pool to allow the pure water to flow out slowly until the seawater samples completely replace the pure water. Ring system (subject to the stability of the curve value within 0.005 m<sup>-1</sup>), at this time close the buffer tank switch and stop the injection of seawater samples into the buffer tank, record the ACS measurement curve and the temperature and salt value at this time are used for data processing, and then take a sample of seawater mixed in the middle of the buffer tank and add it to a 1 cm cuvette. PE35 is measured. After the measurement of a group of samples is completed, turn on the calibration cylinder switch, use Milli-Q water to flush the system, and start to measure the next group of seawater sample. In order to ensure the stability of ACS measurement, remove the  $c$  tube of ACS for cleaning every 5 sets of data and continue to measure seawater samples after calibration with pure water.

3 Results and analysis

3.1 Evaluation of attenuation measurement accuracy of spectrophotometer

In this paper, the uncertainty of PE35 measurement is analyzed using standard particles with different particle sizes, and the uncertainty is defined as

$$E(\lambda) = \frac{c_m(\lambda) - c_{calc}(\lambda)}{c_m(\lambda)} \times 100\%, \tag{4}$$

Where  $c_m(\lambda)$  is the measured attenuation coefficient of PE35,  $c_{calc}(\lambda)$  is the calculated value of the sample Mie scattering theory.  $E(\lambda)$  is a positive value means that the measured value of PE35 is Overestimated, negative value means underestimated.

Figure 2 is the comparison curve between the measured value of standard particles with different particle sizes and the calculated value of Mie scattering theory and the uncertainty analysis result of the line is obtained by normalizing the theoretical calculated value of the standard particulate matter to the measured particle number concentration. In the 400-800 nm band, the PE35 measurement result The theoretical value of Mie scattering is consistent in the spectral shape, but due to the influence of forward scattering cannot be completely eliminated, the measurement results of PE35 are generally low. The blue error line in the figure shows that the spectral shape of the measurement error fluctuating with the wavelength is consistent with the spectral shape of the calculated value of the standard particulate matter. The difference between the measured value and the calculated value of the standard particulate matter in the peak range is much larger than the range of the valley (when the particle size  $D = 2 \mu m$ , the percentage fixed degree is about -2%; when the particle size  $D = 5 \mu m$ , the uncertainty of the peak and trough is about -3%; when the particle size  $D = 10 \mu m$ , the uncertainty of the peak and trough is about -4%; when the particle size  $D = 10 \mu m$ , the uncertainty of peaks and troughs is about -6%). According to the Mie scattering theory, with the increase of the measured spherical particle size

Figure 2 Comparison of the measured value of the attenuation coefficient of the standard particles with different particle sizes and the theoretically calculated value (normalized by the number concentration of the standard particles, t  
Fig.2 Comparison of the attenuation coefficient measurements and theoretical calculations of the standard particles with different sizes  
(normalized by particle concentration with unit of  $g / m^3$ )

0401004- 4

Page 5

#### Light learn learn Report

In addition, the forward scattering of the suspension accounts for a larger proportion of the total scattering. Strong forward scattering causes the energy received by the beam receiver of the PE35 to increase. The attenuation value of is reduced, and as the particle size increases, the measurement uncertainty of the spectrophotometer increases, which are about -5%, -6%, -8%, -10%.

Take the characteristic bands 412, 420, 443, 490, 531, 550, 555, 660, 670, and 678 nm for the calculated and measured values of 4 standard particles Ratio analysis. The result is shown in Figure 3 (a), the diagonal line is 1: 1, the measured value of PE35 is relatively low relative to the calculated value (approximately true value), and the measured value The size of the error has no obvious relationship with the band. From the point of view of the attenuation coefficient, the band fitting effect is better at the lower value of the actual measured attenuation coefficient. At the measured high value of the attenuation coefficient.

Figure 3 (b) shows the measurement uncertainty distribution, most of which are within -1% ~ -10%, and some bands are located at -10% ~ -15% (412, 670, 678 nm Office). Since the particle size distribution and refractive index parameters of spherical standard particles vary by an average of  $\pm 1.1\%$ , the effect on the results can be known from the calculation The average is  $\pm 0.2\%$ , assuming that the system error caused by humans during the measurement of the instrument has an impact of  $\pm 2\%$  on the results, indicating the uncertainty of PE35 About -8% (when the measured particle size is 20  $\mu m$ ), when the measured particle size is as small as 2  $\mu m$ , it is about -3%.

Figure 3 Analysis of standard particulate matter in different characteristic bands. (a) Distribution of measured and calculated values of PE35; (b) Uncertainty distribution of measured values of PE35  
Fig.3 Analysis of the standard particles with different characteristic bands. (A) Distribution of measurement from PE35 and simulation;  
(b) uncertainty distribution of measurement from PE35

### 3.2 Uncertainty analysis of ACS attenuation measurement

Due to the large sample size required by the ACS laboratory cycle measurement system, it is not suitable to use standard particulate suspensions to analyze its uncertainty. by The accuracy of PE35 in section 3.1 is known, so the seawater sample is used as the measured sample for synchronous comparison of ACS and spectrophotometer.

Figure 4 shows the comparison results. ACS (25 cm) attenuation coefficient range is 0 ~ 32  $m^{-1}$ , ACS (10 cm) range is 0 ~ 80  $m^{-1}$ , the part that exceeds the range Not shown in the figure. It can be seen from Figure 4 that the measured value of ACS in the 400-800 nm band is lower than that of PE35. Possible reasons for analysis are: Errors caused by human operation during the measurement process; although the c-tube wall of ACS is composed of black light-absorbing material, the tube wall cannot fully absorb the measured sample Scattered photons; the field of view of the optical path receiver of the ACS is 0.9°, which causes part of the forward scattered light to be received by the optical receiver. underestimate.

In low-turbidity water bodies, the measured values of ACS (25 cm) and ACS (10 cm) are slightly different. The measured values of ACS (25 cm) are higher than those of ACS (10 cm) Slightly higher, closer to the measured value of PE35 (approximately true value), this is because when the divergence angle of the light source is fixed, increasing the optical path can not only make the path During the sample propagation, the scattered part is fully absorbed by the black tube wall, and at the same time, the photon receiver receives less forward scattered photons, Thereby improving the measurement accuracy of c value.

From the analysis of the wave band, the uncertainty of ACS measurement is generally higher than that of the long wave band in the short wave band, which is positively correlated with the wavelength. There is no obvious relationship between the degree of certainty and the wavelength. The turbidity of the water body has a greater influence on the measurement accuracy of ACS. Under low turbidity water The value is 10% ( $\pm 2\%$ ) lower, and as low as 20% ( $\pm 2\%$ ) under high turbidity water bodies.

Taking 420,443,531,550,660,678 nm characteristic bands to analyze the measured values of ACS and PE35, as can be seen from Figure 5, the low turbidity Under the water body, the measured value of ACS is closer to the measured value of PE35, but as the value of c increases, the difference between the measured value of ACS and PE35 gradually Increase. The more turbid the water, the more drastically the accuracy of the ACS measurement will drop. This is mainly due to the strong forward scattering of the particles in the highly turbid water, and

Figure 4 Comparison of ACS and PE35 measurement results

Fig.4 Comparison of measurements between ACS and PE35

The receiver received too much energy, and the value of  $c$  was seriously underestimated. Among them, ACS (25 cm) is saturated when measuring high turbidity, so it appears as shown in Figure 5 (a) at  $c$  50 m<sup>-1</sup>.) The error suddenly increases at 4 points.

From the uncertainty distribution of ACS, the comparison results of ACS (10 cm), ACS (25 cm) and PE35 are shown in Figure 5 (b). The measurement error of ACS varies with the turbidity. The increase shows a significant increase trend, and its uncertainty is mainly concentrated at -1% ~ -15%, and a small part is at -15% ~ -20% (420 nm and 678 nm).

Tables 2 and 3 give the average relative error of each band of ACS (optical path 25 cm) and ACS (optical path 10 cm), / - indicates that ACS (25 cm) is out of range, The average relative error of ACS (25 cm) is 4.145%, and the average relative error of ACS (10 cm) is 9.046%. Indicate the measurement of ACS and spectrophotometer

The accuracy is within acceptable limits.

Figure 5 (a) Distribution of measured values of ACS and PE35 in different characteristic bands; (b) Comparison of ACS uncertainty and measured values of PE35  
Fig. 5 (a) Distribution of measurements from ACS and PE35 with different characteristic bands; (b) comparison between ACS uncertainty

and measurements from PE35

Table 2 Relative deviation of the measured value of ACS (25 cm) in the characteristic band

Table 2 Relative deviation of the measurements from ACS (25 cm) on the characteristic bands

Wavelength / nm	Sample_a /%	Sample_b /%	Sample_c /%	Sample_d /%	Sample_e /%	Sample_f /%	Sample_g /%	Sample_h /%
420	10.404	8.637	8.101	/.	/.	/.	/.	/.
443	10.011	9.073	6.509	/.	/.	/.	/.	/.
531	9.514	8.251	4.094	/.	/.	/.	/.	/.
550	9.147	8.074	3.929	/.	/.	/.	/.	/.
660	6.526	7.489	2.015	4.184	/.	/.	/.	/.
678	6.630	7.479	2.018	3.729	/.	/.	/.	/.
Mean	8.705	8.167	4.455	3.957				
Total mean /%				4.145				

Table 3 Relative deviation of ACS (10 cm) measured value in characteristic band

Table 3 Relative deviation of the measurements from ACS (10 cm) on the characteristic bands

Wavelength / nm	Sample_a /%	Sample_b /%	Sample_c /%	Sample_d /%	Sample_e /%	Sample_f /%	Sample_g /%	Sample_h /%
420	10.204	9.018	8.102	9.843	7.842	12.738	16.427	18.991
443	10.420	8.823	7.742	9.451	7.442	11.844	16.956	16.550
531	10.506	7.555	5.482	7.411	6.172	10.086	14.894	11.445
550	10.271	7.507	5.260	6.979	5.917	9.735	14.554	10.736
660	8.061	6.986	3.567	4.567	4.360	7.374	12.413	6.6840
678	8.261	7.122	3.663	4.428	4.302	7.075	12.170	6.3180
Mean	9.620	7.835	5.636	7.113	6.005	9.808	14.569	11.787
Total mean /%				9.046				

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In order to analyze the uncertainty of the underwater hyperspectral attenuation measurement, the measurement value of PE35 is used as a reference, and the measurement system of PE35 and ACS is compared. The comparative analysis of the step measurement results gives the uncertainty of ACS. Passes 4 standard particles with different particle sizes (center particle size  $D = 2, 5, 10, 20 \mu\text{m}$ ). The accuracy of the PE35 measurement is verified, and the results show that the PE35 measurement results and the theoretical values calculated by Mie scattering have a better spectral shape. Consistent, but the measured value of PE 35 is generally low in amplitude. As the particle size increases, the underestimation of PE35 relative to its theoretical value

0401004- 7

#### Light learn learn Report

The increase is about -5%, -6%, -8%, -10%, and the average uncertainty is about -7%. Then through the ACS laboratory flow measurement system and spectrometer photometer performed a synchronous comparative measurement of the seawater samples obtained in the East China Sea voyage in July 2012. The measurement of ACS was based on the measured value. The analysis was carried out with certainty, and the results showed that the measured value of ACS at low turbidity was 10% ( $\pm 2\%$ ) lower than that measured by the spectrophotometer, and it was 20% ( $\pm 2\%$ ). It was found that in low turbidity water bodies, the measured value of ACS (25cm) is closer to the measured value of PE35 than ACS (10cm) (approximately true value), indicating that ACS (25 cm) is more suitable for measurement in low turbidity water bodies than ACS (10 cm), while ACS (10 cm) is suitable for turbid water bodies such as long measured in Eguchi.

The uncertainty of ACS can be roughly estimated: the measurement uncertainty of PE35 is about -8%. Under low turbidity water bodies, the measurement of ACS (25 cm) relative to PE35 The relative uncertainty of the quantity is about -6% ( $\pm 2\%$ ), so the measurement uncertainty of ACS (25 cm) is about -7.84% ~ -15.36%, ACS (10 cm) is relative to PE35 The relative uncertainty of the measurement is about -10% ( $\pm 2\%$ ), so the measurement uncertainty of ACS (10 cm) is about -17.2% ~ -19.04%; under high turbidity water body ACS The measurement uncertainty of (10 cm) relative to PE35 is about -20% ( $\pm 2\%$ ), then the uncertainty of ACS (10 cm) is about -26.4% ~ -28.24%.

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